

The non rigid rotator

①c

In rigid rotator concept predicts

that the spectral lines are separated by $2B$.

i.e., $\bar{\nu}_{J \rightarrow J+1} = 2B(J+1) \text{ cm}^{-1}$

But- in some cases (eg, HF), the constant separation $2B$ is not observed. There is a decrease in the separation is observed in the case of HF.

(The separation between successive spectral lines is not a constant).

The internuclear distance (r) increases with J .

This means that the bond-length increases

with J . \therefore The assumption of a rigid bond is only an approximation.

When the bond is elastic, the bond will stretch and compress periodically with a fundamental frequency depending on the masses of the atoms and the elasticity of the bond.

If the motion is simple harmonic, the force constant is given by

$$k = 4\pi^2 \bar{\omega}^2 \mu$$

The variation of B with J is determined by the force constant.

Elasticity of the bond also affects the quantities γ and B during a vibration.

(2)c

The measured B value is an average value over time.

$$B = \frac{h}{8\pi^2 c I} = \frac{h}{8\pi^2 c \mu r^2}$$

$$\therefore B \propto \frac{1}{r^2}$$

The average value (over time) of $\frac{1}{r^2}$ is not equal to $\frac{1}{r_e^2}$, where r_e is the equilibrium distance between the atoms.

Consider a bond with equilibrium separation r_e , of 0.1 nm vibrating between the limits 0.09 and 0.11 nm .

$$\langle r \rangle_{av} = \frac{0.09 + 0.11}{2} = 0.1 = r_e$$

$$\text{But } \left\langle \frac{1}{r^2} \right\rangle_{av} = \frac{\left(\frac{1}{0.09} \right)^2 + \left(\frac{1}{0.11} \right)^2}{2} = 103.05 \text{ nm}^{-2}$$

$$\text{and } \langle r \rangle_{av} = \sqrt{\frac{I}{103.05}} = 0.0985 \text{ nm}$$

Hence usually 3 different B values are possible. At equilibrium separation r_e , $B = B_e$.

In the vibrational ground state the average inter nuclear separation r_0 is associated with $B = B_0$. At any level the inter nuclear distance r_v has $B = B_v$.

Spectrum of non rigid rotator

(1) d

Using Schrodinger equation, the rotational energy levels of a non-rigid rotator can be calculated using,

$$E_J = \frac{h^2}{8\pi^2 I} J(J+1) - \frac{h^4}{32\pi^4 I^2 r^2 k} J^2(J+1)^2$$

$$E_J = \frac{E_J}{hc} = B J(J+1) - DJ^2(J+1)^2 \quad \text{Joule} \quad \text{cm}^{-1}$$

$B \rightarrow$ rotational constant

$D \rightarrow$ centrifugal distortion constant

\hookrightarrow harmonic approximation

$$D = \frac{h^3}{32\pi^4 I^2 r^2 k} \quad \text{cm}^{-1}$$

For anharmonic oscillations,

$$\begin{aligned} I &= Mr^2 \\ \mu &= \frac{I}{r^2} \\ \mu &= \frac{Mr^2}{r^2} \\ \mu &= \frac{M}{r} \end{aligned}$$

$$E_J = B J(J+1) - DJ^2(J+1)^2 + H J^3(J+1)^3 + k J^4(J+1)^4 \dots \text{cm}^{-1}$$

From the expression for B and D ,

$$D = \frac{16 B^3 \pi^2 M c^2}{k}$$

$$\begin{aligned} B^3 c^2 &= \frac{h^3}{512\pi^8} \\ D &= \frac{16 B^3}{c^2} \\ &= \frac{16 \cdot \frac{h^3}{512\pi^8}}{c^2} \\ &= \frac{h^3}{32\pi^8 I^2 r^2 k} \end{aligned}$$

$$\begin{aligned} B &= \frac{h^2}{8\pi^2 I c} \\ B^3 &= \frac{h^3}{8\pi^2 I^3 c^3} \\ B^3 c^2 &= \frac{h^3}{512\pi^8 I^3 c} \\ &= \frac{h^3}{512\pi^8} \cdot \frac{1}{M^3 r^3} \end{aligned}$$

D → σ^- $h\nu$ order σ^- 10^{-3} cm^{-1}

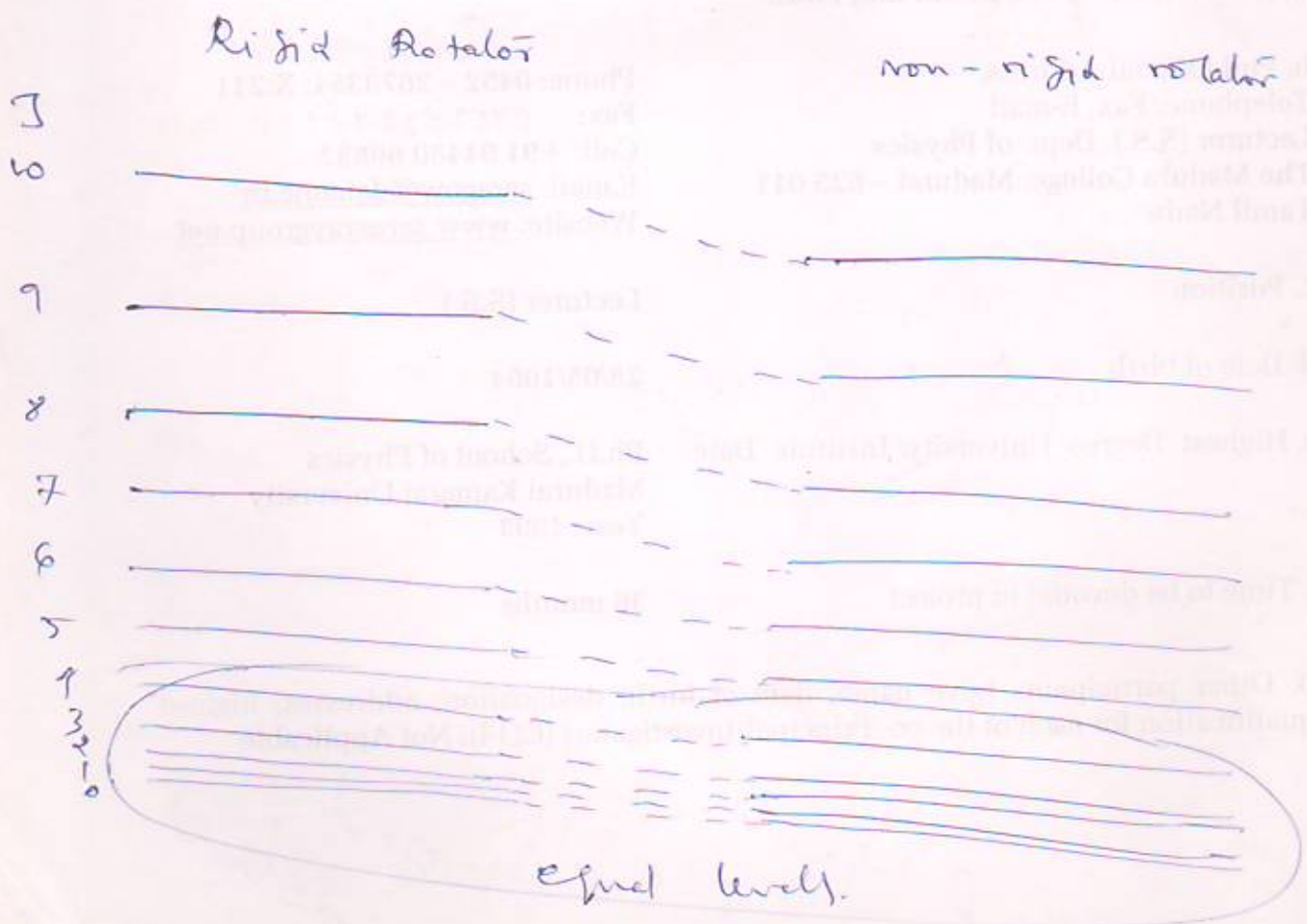
B → σ^- 10 cm^{-1}

∴ D is negligible compared to B.

∴ $D(J+1)^2 J^2$ is also negligible, for small J.

For larger J values ~~D~~ the correction $D J^2 (J+1)^2$ becomes appreciable.

The lowering of energy levels when going from rigid to non-rigid rotation is shown in the figure.



As J increases, D becomes appreciable and energy is lowered in non-rigid rotator
 $E(J) = B J(J+1) - D J^2 (J+1)^2$

$$E_{J+1} - E_J = \bar{V}_J$$

$$= B [(J+1)(J+2) - J(J+1)]$$

$$- D [(J+1)^2 (J+2)^2 - J^2 (J+1)^2]$$

~~$$= B [J^2 + 2J + J + 2 - J^2 - J]$$~~

~~$$= D J$$~~

$$= B [(J+1) \{ (J+2) - J \}]$$

$$- D [(J+1)^2 \left\{ \frac{(J+2)^2 - J^2}{(J^2+1+4J-J^2)} \right\}]$$

$$= 2B (J+1) - D [J^2+1+2J (4+4J)]$$

$$= 2B (J+1) - D [4J^2+4+8J+4J^3+4J+8J^2]$$

$$= \dots - 4D [J^3+1+2J+J^3+J+2J^2]$$

$$= \dots - 4D [J^3+3J^2+3J+1]$$

$$\bar{V}_J = 2B (J+1) - 4D (J+1)^3 \text{ cm}^{-1}$$

From D, the $\bar{\omega}$ values can be determined.

$$\bar{\omega}^2 = \frac{4B^3}{D}$$

A hence $h \bar{\nu}$ for a constant

$k = 4\pi^2 c^2 \bar{\omega}^2 \mu$ can also be determined.